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# Electrical Analysis of Combination of Orange Peel and Tamarind for Bio-battery Application as an Alternative Energy

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# ABSTRACTS

The use of disposable primary batteries causes environmental pollution due to the lack of battery waste treatment facilities, so it is necessary to handle the problem by making bio-batteries from orange peel and tamarind. The purpose of this study is to know the influence of good electricity on variations of the combination of orange peel and tamarind. The novelty of this study is (1) the use of biomass combination of orange peel and tamarind as biobattery electrolytes and (2) comparison of orange peel to tamarind. The method used in this study is by experimenting with a mixture of sweet orange peel and tamarind made into pasta and put in a used battery. The results showed that biobatteries with a variation of 25% orange to tamarind had the highest voltage and the most powerful electric current. This happens because the ion content in the variation is not concentrated so it is easy to move widely. In conclusion, biobatteries with a mixture of orange peel and acid are the best variation of 25% orange peel against tamarind. The study is expected to reduce the use of chemicals in batteries and reduce disposable battery waste.

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## **1. INTRODUCTION**

The use of disposable primary batteries can cause pollution to the environment due to the lack of means of treating waste from used batteries so that they end up in the trash (Widyaningsih & Mulud, 2017). One of the efforts to reduce this pollution is by making bio-batteries (Ratnawati, *et al.*, 2010). Bio-battery is a device that generates electrical energy from materials derived from living things. Fruits are among the ingredients with the greatest potential to be used as bio-batteries (Fauzia, *et al.*, 2019). Orange peels (Salafa, *et al.*, 2020) and tamarind (Suciyati & Supriyanto, 2019) have high electrolyte content, so they can be used as electrolytes in batteries.

Several studies on bio-batteries have been conducted including bio-batteries from *Durio zibenthinus* peels (Febriyanto, *et al.*, 2019), coconut pulp (Zhang, *et al.*, 2016), bananas (Hotang, *et al.*, 2018), *Terminalia cattapa* (Yan & Saha, 2012), as well as oranges and orange peels (Fauzia, et al., 2019). However, there has not been found a study using a combination of orange peel and tamarind as an electrolyte in the bio-battery.

The purpose of this study was to determine the effect of good electricity on the variation of the combination of orange peel and tamarind for bio-battery. The method used is to crush the orange peel and tamarind until it becomes a paste, mix it according to the variations, and put it in a used battery. The novelty of this research are (1) the use of a combination of biomass of orange peel and tamarind as bio-battery electrolyte and (2) the ratio of orange peel to tamarind.

## 2. METHODS

Figure 1 is the process of making a bio-battery using an experimental method. The materials used in this research were orange peel, tamarind, carbon sticks, and used 1.5 volt batteries. Orange peel and tamarind crushed into a paste. After that, the two ingredients are mixed. The variations in the combination of orange peel against tamarind used were 0, 25, 50, 75, and 100%. Then, put into a used battery that has been cleaned.

#### **3. RESULTS AND DISCUSSION**

#### 3.1. Voltage

**Figure 2** is the result of the voltage test from the variation of orange peel and tamarind ratio. The test results showed that the highest voltage was obtained by the variation of orange peel with a percentage of 25%, namely 1.7 volts, while the lowest was obtained from variations in orange peels with a percentage of 100%, namely 1.43 volt. Changes in the variation of the peel of orange and tamarind affect the voltage (Erviana, *et al.*, 2020) because the ions gather in adhering to the cathode so that the amount of charge flows from anode to cathode. The more ions collected, the higher the voltage generated (Atina, 2015).

#### 3.2. Electric Current

Based on the results of observations of electric currents from the variation in the percentage of orange peel to tamarind shown in **Table 1.** The results showed that the highest value of electric current was owned by the variation of orange peel with a percentage of 25%, namely 1.33 mA, while the lowest value was owned by the variation of orange peel with a percentage of 0%, namely 0.63. In the variation of 0% orange peel has the meaning that the content of 100% tamarind that indicates variations of tamarind in the paste very

concentrated, this density results in the content of ions produced very tight and crowded so that the existing ions are difficult to move and produce a small electric current, while in the variation of 25% of existing ions have a wide movement so that it is easy to move because of the decrease in density obtained from orange peel (Fadillah, 2019). The value of this current is influenced by the amount of voltage and resistance in the battery (Abidin, *et al.*, 2020).

#### 4. CONCLUSION

The ratio of orange peel to tamarind has an influence on the magnitude of the voltage and the strength of the electric current. The best bio-battery is when the percentage of orange peel to tamarind was 25%.







Figure 2. Voltage of percentage of orange peel against tamarind

**Table 1.** Electric current from the varitation in the percentage of orange peel againsttamarind

| Percentage of orange peel<br>against tamarind (%) | Electric Current (mA) |
|---|-----------------------|
| 0   | 0.62                  |
| 25  | 1.33                  |
| 50  | 0.92                  |
| 75  | 0.95                  |
| 100   | 1.20                  |

#### 5. AUTHORS' NOTE

The author states that there is no conflict of interest regarding the publication of this article. The author confirms that the paper is free from plagiarism. There are still many shortcomings and need further research from this article so that the resulting battery is used like a conventional battery.

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